

2D MANGANESE(II) NETWORKS BASED ON V-SHAPED BIS(*P*-CARBOXYPHENYL)DIPHENYLSILANE: SYNTHESIS, STRUCTURES AND PROPERTIES

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Two coordination networks based on V-shaped bis(*p*-carboxyphenyl)diphenylsilane (cpdps) ligand and Mn(II) ions: $\{[\text{Mn}_5\text{L}(\text{cpdps})_4(\text{HCOO})_2(\text{H}_2\text{O})_2(\text{DMF})_4] \cdot 6\text{DMF}\}_n$ (**I**) and $\{\text{Mn}_3(\mu\text{-H}_2\text{O})_2(\text{Hcpdps})_2(\text{cpdps})_2(\text{DMF})_2\} \cdot 2\text{DMF} \cdot 12\text{H}_2\text{O}\}_n$ (**II**) were synthesized by conventional and solvothermal methods and structurally characterized by X-ray single crystal diffraction. The pentanuclear and trinuclear manganese clusters in **I** and **II**, respectively, act as SBUs, while the V-shaped ligand links the SBUs leading to 2D coordination networks (Fig. 1).

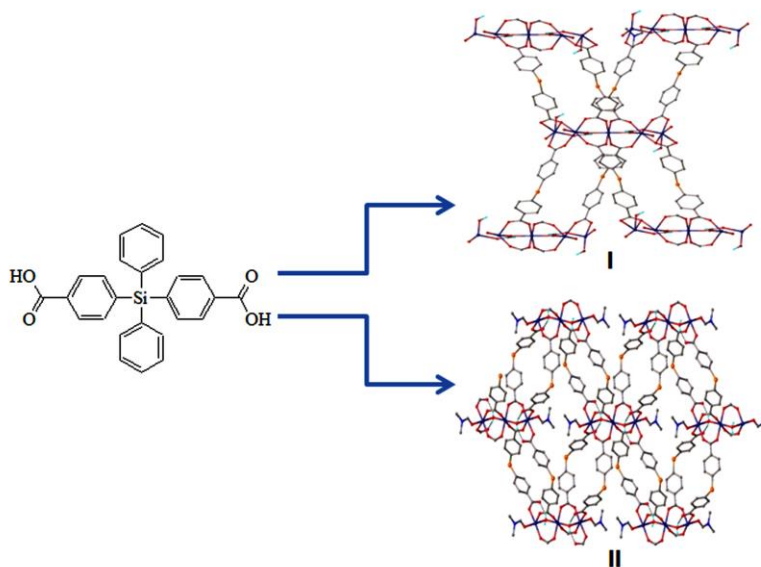


Figure 1. Obtaining of the Mn(II) coordination networks by conventional (**I**) and solvothermal (**II**) methods

Magnetic susceptibility measurements in the 0-300 K range of both coordination networks show the existence of anti-ferromagnetic exchange interactions between the Mn(II) ions. The thermogravimetric data revealed a good thermal stability of both complexes, while the results of the water vapor sorption capacity in dynamic regime indicated good moisture stability of the manganese(II) coordination networks

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